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Key indicators

Single-crystal X-ray study
 $T = 292\text{ K}$
 Mean $\sigma(\text{Se}-\text{As}) = 0.002\text{ \AA}$
 R factor = 0.037
 wR factor = 0.099
 Data-to-parameter ratio = 26.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Tetracaesium octaselenidotetraarsenate(III)

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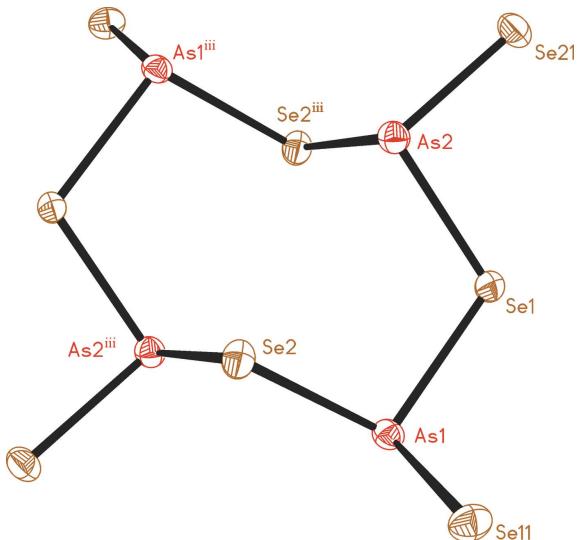
The title compound, $\text{Cs}_4\text{As}_4\text{Se}_8$, contains discrete cyclic $[\text{As}_4\text{Se}_8]^{4-}$ anions consisting of corner-sharing AsSe_3 pyramids. A twist-chair conformation is observed for the eight-membered As_4Se_4 rings, which exhibit a crystallographic inversion centre. Both of the crystallographically independent Cs^+ cations contain eight Se atoms in their coordination spheres.

Comment

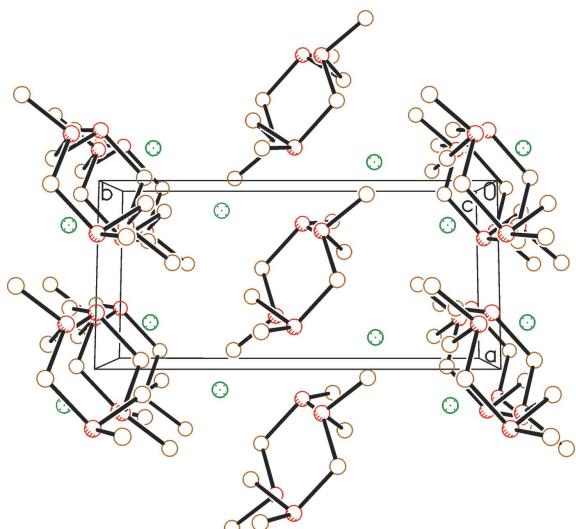
In contrast with the rich structural diversity exhibited by both thioarsenates(III) and thio- and selenidoantimonates(III) (Sheldrick & Wachhold, 1998; Sheldrick, 2000), only a very limited number of structural types are known for the corresponding oligo- and polymeric selenidoarsenates(III). These all contain AsSe_3 pyramids as building units, which are typically connected by shared corners or Se—Se bonds into discrete cyclic anions or infinite chains. The *trans*- $[\text{As}_2\text{Se}_4]^{2-}$ anions of $\text{Ba}_2\text{As}_2\text{Se}_5$ (Cordier *et al.*, 1985) provide the only reported example of edge-sharing. Although cyclic $[\text{As}_3\text{Se}_6]^{3-}$ anions comprising three corner-sharing AsSe_3 pyramids have been structurally characterized in both $[\text{Sr}(\text{en})_4]_2[\text{As}_3\text{Se}_6]\text{Cl}$ (en is ethylenediamine; Sheldrick & Kaub, 1985a) and $[\text{Mn}(\text{dien})_2]_3[\text{As}_3\text{Se}_6]_2$ (dien is diethylenetriamine; Fu, Guo, Liu *et al.*, 2005), many more examples containing $[\text{As}_2\text{Se}_6]^{2-}$ have been isolated in the presence of large counter-cations. Recent structural studies of this chair-shaped dimeric anion with its two Se—Se-linked AsSe_3 pyramids have been reported for $[\text{Cs}(18\text{-crown-6})_2][\text{As}_2\text{Se}_6]$ (18-crown-6 is 1,4,7,10,13,16-hexaoxacyclododecane; van Almsick & Sheldrick, 2005) and $[\text{M}(\text{en})_3][\text{As}_2\text{Se}_6]$ ($\text{M} = \text{Mn}$ or Fe ; Fu, Guo, Liu *et al.*, 2005). The known polymeric selenidoarsenates(III) are restricted to the phases $A\text{AsSe}_2$ [$A = \text{Na}$ (Eisenmann & Schäfer, 1979) and $A = \text{K}-\text{Cs}$ (Sheldrick & Häusler, 1988)] and $A\text{AsSe}_{3-x}\text{H}_2\text{O}$ ($A = \text{K}$ and $x = 1$, or $A = \text{Rb}$ or Cs and $x = 0.5$; Sheldrick & Kaub, 1986). Whereas the $[\text{AsSe}_2]^-$ chains of the former metaselenidoarsenates(III) contain corner-bridged AsSe_3 pyramids, these building units are connected by Se—Se bonds in the $[\text{AsSe}_3]^-$ chains of $A\text{AsSe}_{3-x}\text{H}_2\text{O}$.

The phases $A\text{AsSe}_{3-x}\text{H}_2\text{O}$ were prepared from $A_2\text{CO}_3$ and As_2Se_3 under mild hydrothermal conditions (408 K). We have now observed that the title octaselenidotetraarsenate(III), $\text{Cs}_4\text{As}_4\text{Se}_8$, (I), can be isolated in low yield on heating $\text{CsAsSe}_3 \cdot 0.5\text{H}_2\text{O}$ to 463 K in methanol.

Compound (I) contains the novel crystallographically centrosymmetric $\text{As}_4\text{Se}_8^{4-}$ anion with its eight-membered As_4Se_4 ring (Fig. 1). As observed for the analogous $\text{Sb}_4\text{S}_8^{4-}$ anion of $[\text{Ni}(\text{dien})_2][\text{Sb}_4\text{S}_8]$ (Bensch *et al.*, 2001), the octaselenidotetraarsenate(III) anion exhibits a twist-chair conformation. In accordance with other chalcogenidoarsen-

**Figure 1**

The cyclic $[As_4Se_8]^{4-}$ anion of (I), showing the atom-labelling scheme. Displacement parameters are drawn at the 50% probability level. [Symmetry code: (iii) $1 - x, 1 - y, 1 - z$.]

**Figure 2**

A projection of the structure of (I) perpendicular to [001]. Cs atoms are indicated by green dotted circles, Se atoms by open brown circles and As atoms by red semi-hatched circles

ates(III) (Sheldrick & Wachhold, 1998), the more polar terminal As—Se distances [2.3092 (14) and 2.3223 (13) Å] are significantly shorter than the bridging As—Se bond lengths [2.4306 (13)–2.4676 (13) Å] within the eight-membered ring. Although the smaller $[As_3Se_6]^{3-}$ anion has been structurally characterized in two salts (Sheldrick & Kaub, 1985b; Fu, Guo, Cai *et al.*, 2005), the analogous $[As_4Se_8]^{4-}$ anion has yet to be reported.

Fig. 2 depicts a projection of the crystal structure of (I) perpendicular to [001]. Cs—Se distances in the range 3.6352 (14)–4.1289 (18) Å are observed for the eightfold coordinated counter-cations Cs1 and Cs2.

Experimental

$CsAsSe_3 \cdot 0.5H_2O$ (300.0 mg, 0.66 mmol) (Sheldrick & Kaub, 1986) was heated to 463 K in methanol (0.8 ml) in a sealed glass tube. After 30 h, the contents were allowed to cool to room temperature at 2.1 K h^{-1} to afford red crystals of (I) in 10% yield, together with black $Cs_2As_3Se_6$ (van Almsick & Sheldrick, 2006) in 32% yield.

Crystal data

$Cs_4As_4Se_8$	$D_x = 4.468 \text{ Mg m}^{-3}$
$M_r = 1463.00$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 24 reflections
$a = 7.1572 (13) \text{ \AA}$	$\theta = 5.1\text{--}17.3^\circ$
$b = 14.862 (3) \text{ \AA}$	$\mu = 26.06 \text{ mm}^{-1}$
$c = 10.515 (4) \text{ \AA}$	$T = 292 (2) \text{ K}$
$\beta = 103.51 (2)^\circ$	Prism, red
$V = 1087.5 (5) \text{ \AA}^3$	$0.12 \times 0.11 \times 0.09 \text{ mm}$
$Z = 2$	

Data collection

Siemens P4 four-circle diffractometer	1594 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.037$
Absorption correction: ψ scan (<i>XPREP</i> in <i>SHELXTL</i> ; Sheldrick, 1995)	$\theta_{\text{max}} = 25.0^\circ$
	$h = -1 \rightarrow 8$
	$k = -1 \rightarrow 17$
	$l = -12 \rightarrow 12$
2600 measured reflections	3 standard reflections every 97 reflections
1916 independent reflections	intensity decay: 0.1%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0697P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.099$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 1.28 \text{ e \AA}^{-3}$
1916 reflections	$\Delta\rho_{\text{min}} = -1.23 \text{ e \AA}^{-3}$
73 parameters	

Table 1

Selected geometric parameters (Å, °).

$Cs1-Se1^i$	3.6877 (13)	$Cs2-Se21^{ix}$	3.8412 (14)
$Cs1-Se11^{ii}$	3.6966 (13)	$Cs2-Se1$	3.8778 (14)
$Cs1-Se2^{iii}$	3.7185 (13)	$Cs2-Se1^{vii}$	3.9099 (16)
$Cs1-Se1$	3.7192 (16)	$Cs2-Se2$	4.1289 (18)
$Cs1-Se21^i$	3.7518 (14)	$Se1-As2$	2.4306 (13)
$Cs1-Se21^{iv}$	3.8114 (16)	$Se1-As1$	2.4676 (13)
$Cs1-Se2^y$	3.8516 (13)	$Se21-As2$	2.3223 (13)
$Cs1-Se11^v$	3.8949 (16)	$Se2-As1$	2.4372 (13)
$Cs2-Se11^{vi}$	3.6352 (14)	$Se2-As2^{iii}$	2.4403 (15)
$Cs2-Se11^{vii}$	3.6801 (14)	$Se11-As1$	2.3092 (14)
$Cs2-Se21^{viii}$	3.7009 (16)	$As2-Se2^{iii}$	2.4403 (15)
$Cs2-Se11$	3.7377 (14)		
$As2-Se1-As1$	108.33 (5)	$Se2-As1-Se1$	101.59 (5)
$As1-Se2-As2^{iii}$	94.98 (5)	$Se21-As2-Se1$	96.01 (5)
$Se11-As1-Se2$	98.49 (5)	$Se21-As2-Se2^{iii}$	101.09 (5)
$Se11-As1-Se1$	104.95 (5)	$Se1-As2-Se2^{iii}$	99.18 (5)
$As2^{iii}-Se2-As1-Se11$	-155.96 (5)	$As2-Se1-As1-Se2$	-8.79 (6)
$As2^{iii}-Se2-As1-Se1$	96.77 (5)	$As1-Se1-As2-Se21$	178.95 (5)
$As2-Se1-As1-Se11$	-110.96 (6)	$As1-Se1-As2-Se2^{iii}$	-78.76 (6)

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $x + 1, y, z$; (v) $-x + 2, -y + 1, -z + 1$; (vi) $x - 1, y, z$; (vii) $-x + 1, -y + 1, -z$; (viii) $-x, -y + 1, -z$; (ix) $-x - \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

The highest peak in the final difference Fourier synthesis is situated 1.84 Å from atom Se11 and the deepest hole 1.76 Å from atom Cs1.

Data collection: *R3m/V Software* (Siemens, 1989); cell refinement: *R3m/V Software*; data reduction: *XDISK* (Siemens, 1989); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 1995); software used to prepare material for publication: *SHELXL97*.

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